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10.213-Problem Sets

Solved Problems on

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container holds a mixture of three nonreacting gases: n_1 moles of the first gas with molar specific heat at constant

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volume C 1, and so on. Find the molar specific heat at constant volume of the mixture, in terms of the molar specific heats and quantities of the three separate gases. Concept:-

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PROPERTIES AND OTHER BASIC
CONCEPTS | 5 1. MEASURED
THERMODYNAMIC PROPERTIES AND
OTHER BASIC CONCEPTS 1.1
PRELIMINARY CONCEPTS - THE
LANGUAGE OF THERMODYNAMICS In
order to accurately and precisely discuss
various aspects of thermodynamics, it is
essential to have a well-defined

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vernacular. As such, a list of some foundational concepts and their definitions are shown

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Equilibrium and Stability Criteria:
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Coupled to 1st and 2nd Laws, Fugacity, Activity, Gibbs-Duhem: Problem Set 5

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thermodynamics, we can predict the amount of energy needed to change a system from an equilibrium state to another. For example it will take about

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75 kJ to change 1 kg of air at

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Since different units are used to measure energy in its different forms, solving thermodynamics problems requires a knowledge of the conversion ratios (for example, the ratio for conversion of foot pounds, which are

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units of work energy, into BTUs, which are units of heat energy).

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Thermodynamic Properties 1. If an object has a weight of 10 lbf on the moon, what would the same object weigh on Jupiter? Jupiter 22Moon c ft ft lbf-ft g =75 g =5.4 g =32 sec sec lbf-sec² c moon cmoon Jupiter Jupiter c mg Wg10×32 W = m = = 59.26 lb gg5.4 mg 59.26×75 W = 139 lbf g32 ...

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Thermodynamic Properties

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